

## Letters to the Editor

### Iodoplatination of a triple bond by platinum(IV) complexes: formation of a $\sigma$ -vinyl derivative

S. A. Mitchenko,<sup>a\*</sup> V. P. Ananikov,<sup>a</sup> V. V. Zamashchikov<sup>a</sup>, and Yu. A. Ustynyuk<sup>b</sup>

<sup>a</sup>Institute of Physical Organic Chemistry and Coal Chemistry, Ukrainian National Academy of Sciences, 70 ul. R. Luxembourg, 340114 Donetsk, Ukraine.

Fax: +7 (062 2) 55 3542. E-mail: mitchenk%infou.donetsk.ua@ussr.eu.net

<sup>b</sup>Department of Chemistry, M. V. Lomonosov Moscow State University, Vorob'evy Gory, 119899 Moscow, Russian Federation.

Fax: +7 (095) 939 1854

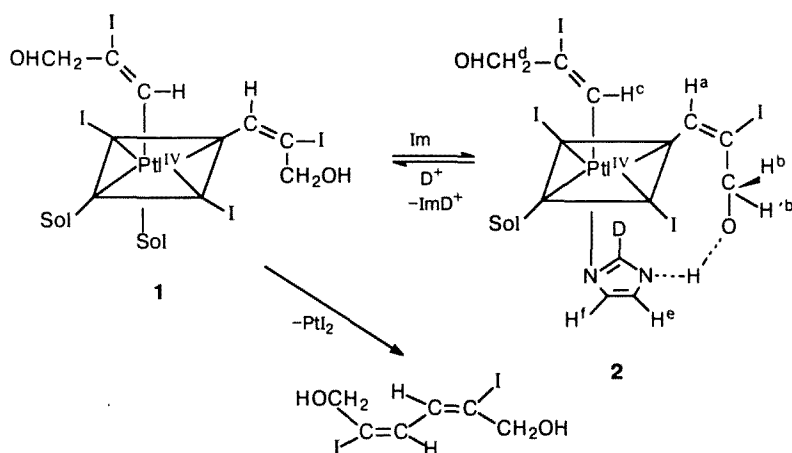
It has been shown previously<sup>1</sup> that hydroiodination of acetylene in aqueous iodide-containing solutions is catalyzed by platinum(IV), presumably *via* intermediate formation of Pt<sup>IV</sup>—CH=CHI  $\sigma$ -vinyl complexes. Using propargyl alcohol as a model compound and in the presence of I<sub>2</sub> added to suppress the possible reduction of Pt<sup>IV</sup> (see Ref. 2), we obtained direct evidence for the occurrence of iodoplatination.

The addition of HC≡CCH<sub>2</sub>OH (78 mg) to the K<sub>2</sub>PtI<sub>6</sub> (426 mg)—KI (125 mg)—I<sub>2</sub> (38 mg)—H<sub>2</sub>O (10 mL) system at 10–15 °C leads to the accumulation of a precipitate over a period of 6 h. Methanol extraction of this precipitate, preliminarily washed with aqueous solution of KI and water, after concentration affords a dark-brown compound (95 g). Found (%): C, 9.8; H, 1.4; I, 61.3; Pt, 23.0. Pt(CH=CI—CH<sub>2</sub>OH)<sub>2</sub>I<sub>2</sub>(CH<sub>3</sub>OH). Calculated (%): C, 9.92; H, 1.42; I, 59.98; Pt, 23.02.

Spectroscopic data confirm the formation of the  $\sigma$ -vinyl derivative: <sup>1</sup>H NMR (CD<sub>3</sub>OD),  $\delta$ : 6.44 (t, 2 H,  $J_{H,H}$  = 2.0 Hz,  $J_{Pt,H}$  = 31.1 Hz); 4.50 (d, 4 H,  $J_{H,H}$  = 2.0 Hz,  $J_{Pt,H}$  = 2.5 Hz); <sup>13</sup>C{<sup>1</sup>H} NMR (CD<sub>3</sub>OD),  $\delta$ : 100.9 (—CH=,  $J_{C,Pt}$  = 801.4 Hz);

82.3 (—CI=,  $J_{C,Pt}$  = 24.4 Hz); 72.6 (—CH<sub>2</sub>—,  $J_{C,Pt}$  = 30.2 Hz). IR (pellets with KBr),  $\nu$ /cm<sup>-1</sup>: 3030 vw  $\nu$ (H—C=), 2920, 2860 w  $\nu$ (CH<sub>2</sub>), 1630 m  $\nu$ (C=O), 1430 w  $\nu$ (CH<sub>2</sub>—C=), 1000, 940 m  $\sigma$ (H—C=).

To elucidate the mutual orientation of the vinyl ligands, 2-deuteroimidazole (Im) was added to a solution of complex **1** obtained in CD<sub>3</sub>OD. Addition of one Im molecule, according to the <sup>1</sup>H NMR data, yields complex **2**, in which the protons of the vinyl ligands are no longer equivalent. As this takes place, three groups of signals appear,  $\delta$ : 7.40 (s, H(a),  $J_{Pt,H(a)}$  = 80.5 Hz); 6.82 (t, H(c),  $J_{cd}$  = 2 Hz,  $J_{Pt,H(c)}$  = 18.2 Hz); 4.62 (d, 2 H(d),  $J_{cd}$  = 2 Hz); 4.02 (dd, H(b) H'(b),  $J_{av,bb'}$  = 3.2 Hz); 7.64 (d, H(f),  $J_{ef}$  = 1.7 Hz,  $J_{Pt,H(f)}$  = 9.0 Hz); 7.24 (d, H(e),  $J_{ef}$  = 1.7 Hz,  $J_{Pt,H(e)}$  = 2.6 Hz) with an H(a) : H(c) : H(f) intensity ratio of 1 : 1 : 1, which corresponds to a *trans*-coordination of Im with respect to one of the vinyl ligands. Taking into account the strong *trans*-effect of vinyl ligands (*cf.* Ref. 3), structure **1** may be attributed to the initial complex, and the nonequivalence of the b and b' protons could be due to the formation of the cyclic structure of complex **2**. The addition of DClO<sub>4</sub> to a solution of complex **2** results in regeneration of the starting compound **1**.



Heating a solution of complex **1** in CD<sub>3</sub>COOD for 3 h at 80 °C affords products of inner-sphere reductive elimination, viz., *E,E*-2,5-diiodohexa-2,4-diene-1,6-diol and PtI<sub>2</sub>. The *E,E* structure of the diene points to the *trans*-addition of platinum(IV) and I<sup>-</sup> to the alkyne in the steps of iodoplatination.

## References

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